

# Synthesis of Cyclopentadiene-Fused Chromanones via One-Pot Multicomponent Reactions

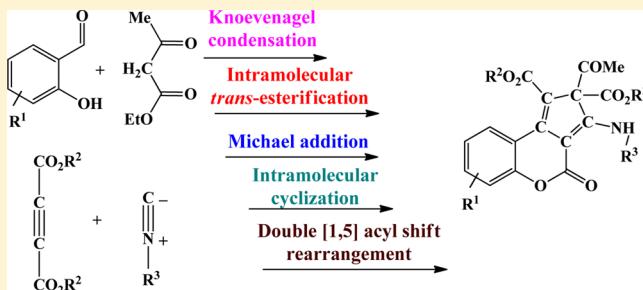
Mehdi Ghandi,<sup>†,\*</sup> Ali-Tabatabaei Ghomi,<sup>†</sup> and Maciej Kubicki<sup>‡</sup>

<sup>†</sup>School of Chemistry, College of Science, University of Tehran, P.O. Box 14155, Tehran 6455, Iran

<sup>‡</sup>Faculty of Chemistry, Adam Mickiewicz University in Poznan, Umultowska 89b, 61-614 Poznan, Poland

## Supporting Information

**ABSTRACT:** We have developed one-pot method for the synthesis of functionalized novel cyclopentadiene-fused chromanone scaffolds. A variety of 4-oxo-2,4-dihydrocyclopenta[c]chromene-1,2-dicarboxylates were obtained in moderate to good yields via condensation of 2-hydroxybenzaldehydes and ethyl acetoacetate with 1:1 acetylenecarboxylate-isocyanides in toluene. These reactions presumably proceed via reaction of the in situ generated 3-acetyl-2H-chromen-2-ones with acetylenecarboxylate-isocyanide zwitterionic intermediates through Michael addition/intramolecular cyclization and double [1,5] acyl shift rearrangement cascade.



## INTRODUCTION

The so-called multicomponent reactions (MCRs) are one-pot processes in which at least three or more different simple substrates react for the preparation of target materials.<sup>1</sup> These reactions, which have gained much attention during the past years, are frequently occurring not through a single-step procedure but rather by several sequential steps or multicomponent cascade or domino reactions.<sup>2</sup> Simplicity, greater efficiency, and atom economy with generation of molecular complexity and diversity in the one-pot transformation are some of the advantages of these reactions. As an important subclass of MCRs, the isocyanide-based multicomponent reactions (IMCRs) are processes in which an isocyanide is used as one of the starting materials in order to obtain new compounds.<sup>3</sup> The pioneering work of Ugi describes the most popular IMCR in which a carboxylic acid, a primary amine, an aldehyde, and an isocyanide react in a one-pot manner to afford an *N*-substituted acyl aminoamide containing four independently varying groups.<sup>4</sup>

The reaction of isocyanides and acetylene compounds, first described by Winterfeld in 1969, is perhaps the founding basis for a large class of new, accessible scaffolds.<sup>5</sup> This reaction initially affords a zwitterionic adduct, which might undergo cycloaddition to activated alkenes, leading to a variety of novel highly substituted cyclopentadienoid systems.<sup>6</sup>

The abundance of naturally occurring chromene and chromane derivatives, and their interesting physiological properties have gained a vital place in the field of heterocyclic chemistry.<sup>7</sup> The chromanones are heterocycles with medicinal properties.<sup>8</sup> Chromone and coumarin derivatives have been found to exhibit a broad range of biological activities such as antiviral,<sup>9</sup> antimicrobial,<sup>10</sup> and antitumor ones.<sup>11</sup> The 3,4-

dihydrocoumarin system is widely distributed in nature, and some derivatives have been shown to exhibit pharmacological activity.<sup>12</sup> On the other hand, the hebertane sesquiterpene heberterenolide, which contains a cyclopentane-fused chromanone, is a biologically active compound.<sup>13</sup>

In continuation of our investigation in searching for MCRs,<sup>14</sup> we decided to explore the idea that trapping of acetylenecarboxylate-isocyanide zwitterionic intermediates with 3-acetyl-2*H*-chromen-2-ones might be hopeful in gaining access to new types of chromanones, annulated-cyclopentadiene moieties, via a three-component reaction. In addition, it seemed intriguing to involve a domino process in this IMCR by increasing the number of reactants if we could carry out the *in situ* generation of 3-acetyl-2*H*-chromen-2-ones in the same pot. Herein, we report the synthesis of novel 4-oxo-2,4-dihydrocyclopenta[c]chromene-1,2-dicarboxylates via a one-pot, four-component reaction of a variety of 2-hydroxybenzaldehydes with ethyl acetoacetate, acetylenedicarboxylates, and cyclohexyl or *tert*-butyl isocyanides in toluene.

## RESULTS AND DISCUSSION

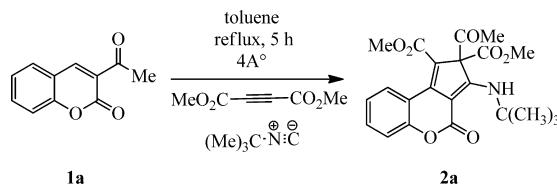
3-Acetyl-2*H*-chromen-2-one (**1a**) served for our early exploration. Indeed, **2a** was isolated as the sole reaction product upon treatment of **1a** with an equimolar amount of dimethyl acetylenedicarboxylate (DMAD) and *tert*-butyl isocyanide within 2 h, in refluxing toluene (Scheme 1).

Our later studies revealed that **2a** could be synthesized in a one-pot reaction if **1a** prepared *in situ* from piperidine-catalyzed condensation of salicylaldehyde (**3a**) with ethyl

Received: December 27, 2012

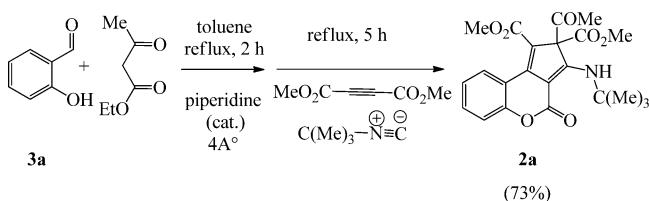
Published: February 6, 2013

**Scheme 1. Formation of 2a from Reaction of 1a with DMAD and *tert*-Butyl Isocyanide**



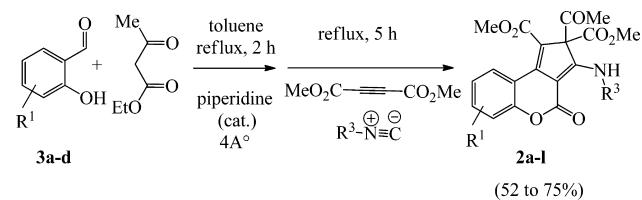
acetoacetate at reflux within 2 h was treated with equimolar amounts of DMAD and *tert*-butyl isocyanide within 3 h under reflux conditions (Scheme 2). Therefore, the implication of 3-acetyl-2*H*-chromen-2-one (**1a**) in the initial reaction is confirmed.

**Scheme 2. One-Pot Synthesis of 2a**



Encouraged by this result, this new method was then applied to a range of 2-hydroxybenzaldehydes **3a–d**, *tert*-butyl, cyclohexyl, or 1,1,3,3-tetramethylbutyl isocyanides and DMAD (Table 1) or diethyl acetylenedicarboxylate (DEAD)

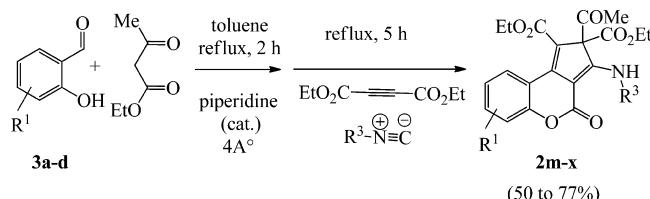
**Table 1.** Products 2a–l Obtained via One-Pot Reaction of 3a–d with DMAD and Isocyanides



entry	R <sup>1</sup>	R <sup>3</sup>	product	yield (%)
1	H	<i>tert</i> -butyl	<b>2a</b>	73
2	H	cyclohexyl	<b>2b</b>	75
3	H	tetramethylbutyl	<b>2c</b>	65
4	S-Br	<i>tert</i> -butyl	<b>2d</b>	74
5	S-Br	cyclohexyl	<b>2e</b>	75
6	S-Br	tetramethylbutyl	<b>2f</b>	64
7	S-NO <sub>2</sub>	<i>tert</i> -butyl	<b>2g</b>	58
8	S-NO <sub>2</sub>	cyclohexyl	<b>2h</b>	54
9	S-NO <sub>2</sub>	tetramethylbutyl	<b>2i</b>	52
10	3-OMe	<i>tert</i> -butyl	<b>2j</b>	59
11	3-OMe	cyclohexyl	<b>2k</b>	63
12	3-OMe	tetramethylbutyl	<b>2l</b>	53

(Table 2). The structures of **2a–p** were deduced by elemental analysis, MS, IR, and  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy. For example, part of the  $^1\text{H}$  NMR spectrum of dimethyl 2-acetyl-8-bromo-3-(cyclohexylamino)-4-oxo-2,4-dihydrocyclopenta[c]-chromene-1,2-dicarboxylate (**2e**) exhibited four singlets at  $\delta$  2.75 (3H), 3.74 (3H), 3.77 (3H), and 9.61 (1H) due to MeCO, MeOCO, MeOCO, and NH, respectively. The  $^1\text{H}$ -decoupled  $^{13}\text{C}$  NMR revealed three characteristic signals at  $\delta$  195.7, 167.1, and 165.0 due to three carbonyl groups. Unambiguous evidence

**Table 2. Products 2m-x Obtained via One-Pot Reaction of 3a-d with DEAD and Isocyanides**



entry	R <sup>1</sup>	R <sup>3</sup>	product	yield (%)
1	H	<i>tert</i> -butyl	<b>2m</b>	72
2	H	cyclohexyl	<b>2n</b>	73
3	H	tetramethylbutyl	<b>2o</b>	50
4	5-Br	<i>tert</i> -butyl	<b>2p</b>	77
5	5-Br	cyclohexyl	<b>2q</b>	72
6	5-Br	tetramethylbutyl	<b>2r</b>	62
7	5-NO <sub>2</sub>	<i>tert</i> -butyl	<b>2s</b>	55
8	5-NO <sub>2</sub>	cyclohexyl	<b>2t</b>	56
9	5-NO <sub>2</sub>	tetramethylbutyl	<b>2u</b>	50
10	3-OMe	<i>tert</i> -butyl	<b>2v</b>	55
11	3-OMe	cyclohexyl	<b>2w</b>	57
12	3-OMe	tetramethylbutyl	<b>2x</b>	54

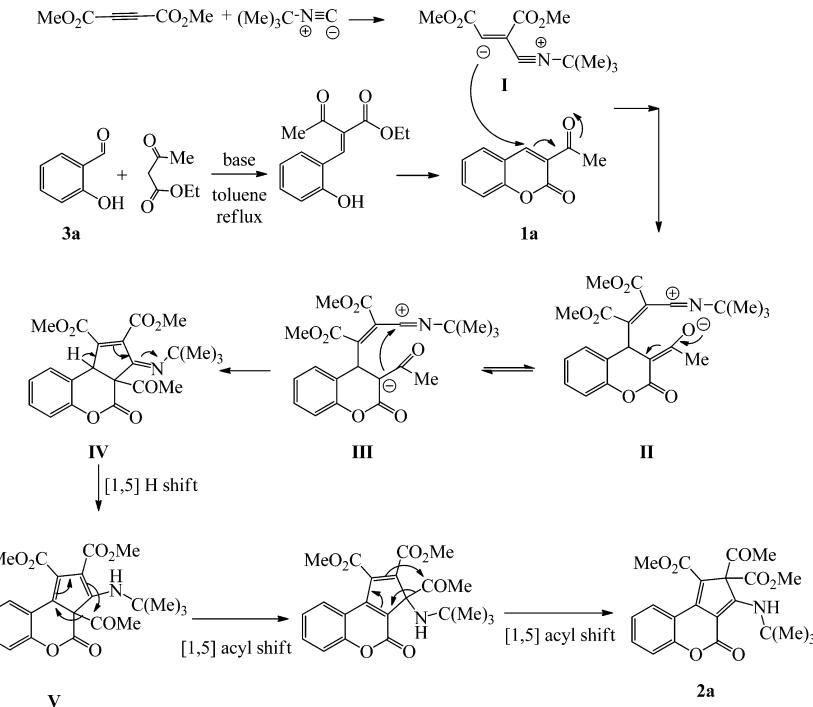
for the proposed structure of **2q** was finally obtained by single-crystal X-ray-diffraction analysis (Figure 1).



**Figure 1.** X-ray crystal structure of compound **2q**.<sup>16</sup>

Although the precise mechanism is not known, a mechanistic postulate as shown in Scheme 3 may be invoked to rationalize the formation of **2a**. It is conceivable that the zwitterionic intermediate **I**, formed by the 1:1 interaction between the isocyanide and acetylenedicarboxylate,<sup>15</sup> attacks preferentially the C-4 of 3-acetyl-2*H*-chromen-2-one (**1a**) leading to 1,7- or 1,5-dipolar intermediates **II** and **III**. The subsequently generated **IV** upon ring closure is transformed into **V** via an allowed [1,5] H shift. It is finally transformed into **2a**, presumably through two consecutive allowed [1,5] acyl shifts. Generation of an extra conjugation between cyclopentadiene ring and six-membered C=O group seems to be the driving force for these rearrangements.

As indicated in Tables 1 and 2, whereas unsubstituted aldehydes (entries 1 to 3) or substituted with a modest electron-withdrawing group (EWG) (entries 4–6) afford products **2a–f** and **2m–r** in higher yields, those bearing a strong EWG (entries 7–9) or a strong electron-donating group (EDG) (entries 10–12) produce the corresponding products **2g–l** and **2s–x** in lower yields. This behavior is the result of a

**Scheme 3.** Mechanistic Rationalization for the Formation of Compound **2a**

multistep reaction in which one step may be rate determining for EWGs, but a different step may become rate limiting for ERG substituents.<sup>17</sup> Although substituted aldehyde with EWG facilitates the formation rate of intermediates **II** and **III** via the addition of zwitterionic intermediate **I** to the double bond of **1a**, the subsequent cyclization step to iminocyclopentene **IV** (Schemes 3) becomes rate limiting for this substituent. In contrast, the first step is expected to be rate limiting for aldehyde bearing EDG. Therefore, obtaining lower yields of **2g–l** and **2s–x** in comparison to other products indicated in Tables 1 and 2 is anticipated. On the other hand, the products **2m–x** (Table 2) have generally been formed in lower yields in comparison to those of **2a–l** (Table 2), perhaps due to larger steric effect of DEAD.

Overall, novel 4-oxo-2,4-dihydrocyclopenta[c]chromene-1,2-dicarboxylates were obtained as the major products in moderate to good yields. More importantly, doing the experiment in one-pot reaction with four different simple substrates is of fundamental importance due to the simplicity, greater efficiency and atom economy with generation of molecular complexity and diversity.

In conclusion, a number of 2-hydroxybenzaldehydes, ethyl acetoacetate, acetylenecarboxylates, and isocyanides underwent one-pot multicomponent reactions in toluene, affording the desired products. These reactions were designed to obtain a variety of biologically interesting cyclopentadiene-fused chromanones. These new structures broaden the scaffolds that are accessible through Knoevenagel condensation—trans esterification—Michael addition—intramolecular cyclization—acyl shift rearrangement cascade and many of them may represent interesting pharmacophores.

## EXPERIMENTAL SECTION

**General Information.** <sup>1</sup>H NMR, <sup>13</sup>C NMR, MS, and elemental analysis were measured with conventional spectrometers. All solvents

were purified and dried by following standard procedures unless otherwise stated.

**Synthesis of Dimethyl 3-(*tert*-Butylamino)-2-acetyl-5-oxacyclopenta[*a*]naphthalen-4(2H)-one-1,2-dicarboxylate Derivatives **2a–x**.** *General Procedure.* To a stirring solution of piperidine (0.17 g, 20 mol %) in toluene (20 mL) containing molecular sieves 4 Å was added 2-hydroxybenzaldehyde derivatives (**3a–d**) (1.0 mmol) and ethyl acetoacetate (1.0 mmol), and the mixture was heated at reflux for 2 h. Acetylenedicarboxylate (1.0 mmol) and isocyanide (1.0 mmol) were then added, and the mixture was heated at reflux for another 5 h. After completion as indicated by TLC, the solvent was removed under reduced pressure and the residue was purified by column chromatography (SiO<sub>2</sub>, eluent: 1:3 *n*-hexane/EtOAc) to afford the products **2a–x**.

**Dimethyl 3-(*tert*-butylamino)-2-acetyl-5-oxacyclopenta[*a*]naphthalen-4(2H)-one-1,2-dicarboxylate (2a):** yellow solid (301 mg, 73%); mp 140–142 °C; IR (KBr)  $\nu_{\text{max}}$  3402, 1711, 1688, 1647, 1603 cm<sup>-1</sup>;  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.66 (9H, s), 2.74 (3H, s), 3.74 (3H, s), 3.77 (3H, s), 7.35–7.43 (2H, m), 7.66–7.89 (2H, m), 9.53 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 25.9, 30.6, 51.3, 52.3, 52.4, 96.2, 107.5, 121.5, 121.8, 125.4, 126.1, 128.7, 129.1, 146.0, 150.1, 155.3, 160.6, 165.0, 167.0, 195.0 ppm; *m/z* (EI, 70 eV) 413 (17, M<sup>+</sup>), 385 (25), 371 (60), 315 (100), 56 (55). Anal. Calcd for C<sub>22</sub>H<sub>23</sub>NO<sub>7</sub>: C, 63.91; H, 5.61; N, 3.39. Found: C, 63.79; H, 5.58; N, 3.33.

**Dimethyl 2-acetyl-3-(cyclohexylamino)-5-oxacyclopenta[*a*]naphthalen-4(2H)-one-1,2-dicarboxylate (2b):** orange solid (329 mg, 75%); mp 150–152 °C; IR (KBr)  $\nu_{\text{max}}$  3408, 1711, 1673, 1638, 1603 cm<sup>-1</sup>;  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.43–2.27 (10H, m), 2.75 (3H, s), 3.49–3.59 (1H, m), 3.74 (3H, s), 3.77 (3H, s), 7.35–7.43 (2H, m), 7.66–7.89 (2H, m), 9.63 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 23.7, 25.2, 26.3, 26.4, 30.0, 30.1, 51.4, 52.3, 52.4, 96.3, 107.5, 121.5, 121.8, 125.4, 126.1, 128.7, 129.1, 146.0, 150.1, 155.3, 160.6, 165.0, 167.0, 195.0 ppm; *m/z* (EI, 70 eV) 440 (15, M<sup>+</sup> + 1), 411 (35), 397 (60), 315 (100), 56 (50). Anal. Calcd for C<sub>24</sub>H<sub>25</sub>NO<sub>7</sub>: C, 65.59; H, 5.73; N, 3.19. Found: C, 65.48; H, 5.78; N, 3.13.

**Dimethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-5-oxacyclopenta[*a*]naphthalen-4(2H)-one-1,2-dicarboxylate (2c):** orange solid (304 mg, 65%); mp 151–152 °C; IR (KBr)  $\nu_{\text{max}}$  3448, 1711, 1688, 1647, 1610 cm<sup>-1</sup>;  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.06 (9H, s), 1.65 (6H, s), 1.78 (2H, s), 2.64 (3H), 3.74 (3H, s), 3.77 (3H, s), 7.35–7.43 (2H, m), 7.66–7.89 (2H, m), 9.53 (1H, s) ppm;  $\delta_{\text{C}}$  (100

MHz,  $\text{CDCl}_3$ ) 25.9, 27.9, 31.0, 31.6, 31.7, 41.4, 52.3, 52.4, 53.4, 96.7, 107.5, 121.5, 121.8, 125.4, 126.1, 128.7, 129.1, 146.0, 150.1, 155.3, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 469 (12,  $M^+$ ), 441 (32), 427 (55), 315 (100), 56 (55). Anal. Calcd for  $\text{C}_{26}\text{H}_{31}\text{NO}_7$ : C, 66.51; H, 6.65; N, 2.98. Found: C, 66.48; H, 6.58; N, 3.03.

**Dimethyl 3-(*tert*-butylamino)-2-acetyl-8-bromo-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2d**):** orange solid (364 mg, 74%); mp 148–150 °C; IR (KBr)  $\nu_{\max}$  3412, 1712, 1701, 1688, 1623  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.66 (9H, s), 2.74 (3H, s), 3.74 (3H, s), 3.77 (3H, s), 7.31 (1H, d,  $J$  8.4 Hz), 7.70 (1H, dd,  $J$  = 8.4, 2.4 Hz), 8.01 (1H, d,  $J$  = 2.4 Hz), 9.60 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 25.9, 30.6, 53.4, 55.3, 59.1, 96.2, 107.5, 118.1, 118.7, 121.1(C<sub>1</sub>), 128.4, 129.1, 132.2, 146.0, 150.1, 155.3, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 494 ( $M^+$  + 2 [<sup>81</sup>Br], 8), 492 ( $M^+$  [<sup>79</sup>Br], 8), 463 (15), 394 (33), 363 (66), 57 (100). Anal. Calcd for  $\text{C}_{22}\text{H}_{22}\text{BrNO}_7$ : C, 53.67; H, 4.50; N, 2.85. Found: C, 53.58; H, 4.48; N, 2.83.

**Dimethyl 2-acetyl-8-bromo-3-(cyclohexylamino)-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2e**):** orange solid (388 mg, 75%); mp 153–155 °C; IR (KBr)  $\nu_{\max}$  3428, 1722, 1711, 1688, 1638  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.43–2.27 (10H, m), 2.75 (3H, s), 3.55 (1H, m), 3.74 (3H, s), 3.77 (3H, s), 7.31 (1H, d,  $J$  8.4 Hz), 7.70 (1H, dd,  $J$  = 8.4, 2.4 Hz), 8.01 (1H, d,  $J$  = 2.4 Hz), 9.61 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 23.8, 25.2, 26.3, 26.4, 30.0, 30.1, 53.2, 55.3, 57.2, 96.3, 107.5, 118.1, 118.7, 121.1, 128.4, 129.1, 132.2, 146.0, 150.1, 155.3, 160.6, 165.0, 167.0, 195.6 ppm;  $m/z$  (EI, 70 eV) 519 ( $M^+$  + 2 [<sup>81</sup>Br], 8), 517 ( $M^+$  [<sup>79</sup>Br], 8), 489 (19), 475 (23), 392 (65), 57 (100). Anal. Calcd for  $\text{C}_{24}\text{H}_{24}\text{BrNO}_7$ : C, 55.61; H, 4.67; N, 2.70. Found: C, 55.58; H, 4.68; N, 2.74.

**Dimethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-8-bromo-5-oxacyclopenta[*a*]naphthalene-4(2*H*)-one-1,2-dicarboxylate (**2f**):** orange solid (350 mg, 64%); mp 160–162 °C; IR (KBr)  $\nu_{\max}$  3458, 1732, 1721, 1688, 1640  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.06 (9H, s), 1.65 (6H, s), 1.78 (2H, s), 2.64 (3H, s), 3.84 (3H, s), 3.90 (3H, s), 7.32 (1H, d,  $J$  8.4 Hz), 7.70 (1H, dd,  $J$  = 8.4, 2.4 Hz), 8.01 (1H, d,  $J$  = 2.4 Hz), 9.60 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 25.8, 27.9, 31.0, 31.6, 31.7, 42.4, 51.2, 53.5, 55.4, 96.4, 107.5, 118.1, 118.7, 121.1, 128.4, 129.1, 132.2, 146.1, 150.0, 155.3, 160.6, 165.0, 167.2, 195.5 ppm;  $m/z$  (EI, 70 eV) 549 ( $M^+$  + 2 [<sup>81</sup>Br], 11), 547 ( $M^+$  [<sup>79</sup>Br], 11), 519 (22), 505 (53), 392 (65), 57 (100). Anal. Calcd for  $\text{C}_{26}\text{H}_{30}\text{BrNO}_7$ : C, 56.94; H, 5.51; N, 2.55. Found: C, 56.97; H, 5.48; N, 2.54.

**Dimethyl 3-(*tert*-butylamino)-2-acetyl-8-nitro-5-oxa-cyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2g**):** yellow solid (265 mg, 58%); mp 158–160 °C; IR (KBr)  $\nu_{\max}$  3402, 1706, 1688, 1622, 1603  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.66 (9H, s), 2.64 (3H, s), 4.04 (3H, s), 4.07 (3H, s), 7.55 (1H, d,  $J$  9.6 Hz), 8.52–8.61 (2H, m), 10.22 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 25.9, 30.6, 53.4, 55.4, 59.1, 96.2, 107.5, 118.7, 121.1, 121.9, 125.4, 128.1, 144.0, 146.0, 153.1, 155.3, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 458 (17,  $M^+$ ), 430 (25), 416 (55), 360 (100), 56 (100). Anal. Calcd for  $\text{C}_{22}\text{H}_{22}\text{N}_2\text{O}_9$ : C, 57.64; H, 4.84; N, 6.11. Found: C, 57.67; H, 4.82; N, 6.12.

**Dimethyl 2-acetyl-3-(cyclohexylamino)-8-nitro-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2h**):** yellow solid (261 mg, 54%); mp 165–167 °C; IR (KBr)  $\nu_{\max}$  3418, 1711, 1701, 1638, 1603  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.43–2.27 (10H, m), 2.75 (3H, s), 3.53 (1H, m), 4.04 (3H, s), 4.07 (3H, s), 7.55 (1H, d,  $J$  9.6 Hz), 8.52–8.61 (2H, m), 10.22 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 23.8, 25.2, 26.3, 30.0, 30.1, 53.2, 55.3, 57.3, 93.2, 107.5, 118.7, 121.1, 121.9, 125.4, 128.1, 144.1, 146.1, 153.1, 155.3, 160.6, 165.1, 167.2, 195.5 ppm;  $m/z$  (EI, 70 eV) 484 (15,  $M^+$ ), 456 (28), 442 (55), 360 (100), 56 (59). Anal. Calcd for  $\text{C}_{24}\text{H}_{24}\text{N}_2\text{O}_9$ : C, 59.50; H, 4.99; N, 5.78. Found: C, 59.48; H, 4.96; N, 5.75.

**Dimethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-8-nitro-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2i**):** yellow solid (267 mg, 52%); mp 166–168 °C; IR (KBr)  $\nu_{\max}$  3428, 1721, 1706, 1647, 1610  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.06 (9H, s), 1.65 (6H, s), 1.78 (2H, s), 2.64 (3H, s), 4.04 (3H, s), 4.07 (3H, s), 7.55 (1H, d,  $J$  9.6 Hz), 8.52–8.61 (2H, m), 10.22 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 25.9, 27.9, 31.0, 31.6, 31.7, 43.4, 51.2, 53.5, 55.4, 96.9, 107.6, 118.7, 121.1, 121.9, 125.4, 128.1, 144.1, 146.1, 153.1, 155.3,

160.6, 165.1, 167.2, 195.4 ppm;  $m/z$  (EI, 70 eV) 514 (18,  $M^+$ ), 486 (45), 472 (65), 416 (100), 56 (55). Anal. Calcd for  $\text{C}_{26}\text{H}_{30}\text{N}_2\text{O}_9$ : C, 60.69; H, 5.88; N, 5.44; N, 5.78. Found: C, 60.65; H, 5.86; N, 5.48.

**Dimethyl 3-(*tert*-butylamino)-2-acetyl-6-methoxy-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2j**):** yellow solid (261 mg, 59%); mp 151 °C; IR (KBr)  $\nu_{\max}$  3402, 1711, 1701, 1647, 1610  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.66 (9H, s), 2.64 (3H, s), 3.74 (3H, s), 3.77 (3H, s), 4.00 (3H, s), 7.28 (1H, d,  $J$  8.4 Hz), 7.71 (1H, dd,  $J$  = 8.4, 2.4 Hz), 8.63 (1H, d,  $J$  = 2.4 Hz), 9.28 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 25.9, 30.6, 52.3, 52.7, 55.7, 59.1, 97.5, 107.5, 113.5, 119.1, 120.1, 125.4, 128.1, 142.1, 147.6, 148.0, 155.3, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 443 (14,  $M^+$ ), 401 (66), 371 (100), 315 (80), 56 (55). Anal. Calcd for  $\text{C}_{23}\text{H}_{25}\text{NO}_8$ : C, 62.30; H, 5.68; N, 3.16. Found: C, 62.27; H, 5.65; N, 3.13.

**Dimethyl 2-acetyl-3-(cyclohexylamino)-6-methoxy-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2k**):** yellow solid (295 mg, 63%); mp 149 °C; IR (KBr)  $\nu_{\max}$  3418, 1711, 1706, 1647, 1610  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.43–2.27 (10H, m), 2.75 (3H, s), 3.53 (1H, m), 3.74 (3H, s), 3.77 (3H, s), 4.02 (3H, s), 7.27 (1H, d,  $J$  8.4 Hz), 7.71 (1H, dd,  $J$  = 8.4, 2.4 Hz), 8.63 (1H, d,  $J$  = 2.4 Hz), 9.26 (1H, s) ppm,  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 23.7, 25.2, 26.3, 26.4, 30.0, 30.1, 52.3, 52.7, 55.7, 57.7, 97.2, 107.4, 113.5, 119.1, 120.1, 125.5, 128.1, 142.1, 147.0, 148.0, 155.4, 160.6, 165.1, 167.1, 195.1 ppm;  $m/z$  (EI, 70 eV) 469 (15,  $M^+$ ), 427 (50), 397 (100), 315 (90), 55 (55). Anal. Calcd for  $\text{C}_{25}\text{H}_{27}\text{NO}_8$ : C, 63.96; H, 5.80; N, 2.98%. Found: C, 63.89; H, 5.75; N, 3.02.

**Dimethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-6-methoxy-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2l**):** yellow solid (295 mg, 63%); mp 147–148 °C; IR (KBr)  $\nu_{\max}$  3428, 1712, 1706, 1647, 1610  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.06 (9H, s), 1.65 (6H, s), 1.78 (2H, s), 2.64 (3H, s), 3.74 (3H, s), 4.03 (3H, s), 7.27 (1H, d,  $J$  8.4 Hz), 7.71 (1H, dd,  $J$  = 8.4, 2.4 Hz), 8.63 (1H, d,  $J$  = 2.4 Hz), 9.26 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 25.8, 27.8, 31.0, 31.6, 31.7, 45.0, 52.7, 53.5, 55.4, 54.7, 97.1, 107.4, 113.5, 119.2, 120.1, 125.5, 128.2, 142.2, 147.0, 148.0, 155.3, 160.6, 165.1, 167.2, 195.3 ppm;  $m/z$  (EI, 70 eV) 499 (15,  $M^+$ ), 471 (40), 427 (35), 315 (100), 57 (95). Anal. Calcd for  $\text{C}_{27}\text{H}_{33}\text{NO}_8$ : C, 64.92; H, 6.66; N, 2.80. Found: C, 64.86; H, 6.68; N, 2.82.

**Diethyl 3-(*tert*-butylamino)-2-acetyl-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2m**):** orange solid (317 mg, 72%); mp 145–147 °C; IR (KBr)  $\nu_{\max}$  3403, 1712, 1688, 1647, 1603  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J$  = 7.2 Hz), 1.30 (3H, t,  $J$  = 7.2 Hz), 1.66 (9H, s), 2.74 (3H, s), 4.13–4.26 (4H, m), 7.35–7.43 (2H, m), 7.66–7.89 (2H, m), 9.60 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 30.6, 51.3, 59.2, 61.4, 95.3, 107.3, 121.1, 121.6, 125.4, 126.1, 128.7, 129.1, 146.0, 150.1, 155.3, 160.6, 165.0, 167.1, 195.3 ppm;  $m/z$  (EI, 70 eV) 441 (17,  $M^+$ ), 385 (21), 371 (40), 315 (100), 56 (55). Anal. Calcd for  $\text{C}_{24}\text{H}_{27}\text{NO}_7$ : C, 65.29; H, 6.16; N, 3.17. Found: C, 65.27; H, 6.12; N, 3.19.

**Diethyl 2-acetyl-3-(cyclohexylamino)-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2n**):** orange solid (340 mg, 73%); mp 148–150 °C; IR (KBr)  $\nu_{\max}$  3418, 1713, 1688, 1648, 1603  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J$  = 7.2 Hz), 1.30 (3H, t,  $J$  = 7.2 Hz), 1.66 (9H, s), 2.75 (3H, s), 3.49–3.59 (1H, m, CHN) 4.13–4.23 (4H, m), 7.35–7.43 (2H, m), 7.66–7.89 (2H, m), 9.60 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 23.7, 25.2, 26.3, 26.4, 30.0, 30.1, 51.4 (CHN, 59.2, 61.4, 95.1, 107.3, 121.1, 121.6, 125.4, 126.1, 128.7, 129.1, 146.1, 150.3, 155.1, 160.0, 165.1, 167.7, 195.1 ppm;  $m/z$  (EI, 70 eV) 467 (15,  $M^+$ ), 411 (28), 397 (65), 315 (100), 56 (55). Anal. Calcd for  $\text{C}_{26}\text{H}_{29}\text{NO}_7$ : C, 66.80; H, 6.20; N, 3.00. Found: C, 66.72; H, 6.23; N, 3.02.

**Diethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-5-oxacyclopenta[*a*]naphthalen-4(2*H*)-one-1,2-dicarboxylate (**2o**):** orange solid (248 mg, 50%); mp 153–154 °C; IR (KBr)  $\nu_{\max}$  3449, 1710, 1700, 1647, 1610  $\text{cm}^{-1}$ ;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.06 (9H, s), 1.17 (3H, t,  $J$  = 7.2 Hz), 1.30 (3H, t,  $J$  = 7.2 Hz), 1.65 (6H, s), 1.78 (2H, s), 2.74 (3H, t,  $J$  = 7.2 Hz), 7.66–7.89 (2H, m), 9.59 (1H, s) ppm;  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 27.9, 31.0, 31.6, 31.7, 41.5, 53.4, 59.2, 61.4, 95.2, 107.6, 121.1, 121.6, 125.4, 126.1, 128.7, 129.1, 146.8, 150.3, 155.2, 160.0, 165.2, 167.6, 195.2 ppm;  $m/z$  (EI, 70 eV) 498 (12,

$M^+ + 1$ ), 427 (15), 343 (50), 315 (100), 57 (57). Anal. Calcd for  $C_{28}H_{33}NO_7$ : C, 67.59; H, 7.09; N, 2.81. Found: C, 67.64; H, 7.02; N, 2.82.

**Diethyl 3-(tert-butylamino)-2-acetyl-8-bromo-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2P):** orange solid (400 mg, 77%); mp 153–155 °C; IR (KBr)  $\nu_{\max}$  3413, 1712, 1701, 1688, 1633  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.66 (9H, s), 2.74 (3H, s), 4.13–4.26 (4H, m), 7.32 (1H, d,  $J = 8.4 \text{ Hz}$ ), 7.70 (1H, dd,  $J = 8.4, 2.4 \text{ Hz}$ ), 8.01 (1H, d,  $J = 2.4 \text{ Hz}$ ), 9.60 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 30.6, 51.3, 59.3, 61.4, 95.5, 107.4, 118.1, 118.7, 121.1, 128.5, 129.1, 132.2, 146.0, 150.0, 155.6, 160.3, 165.0, 167.1, 195.5 ppm;  $m/z$  (EI, 70 eV) 521 ( $M^+ + 2$  [81Br], 14), 519 (M + [79Br], 8), 505 (30), 433 (100), 417 (60), 55 (90). Anal. Calcd for  $C_{24}H_{26}\text{BrNO}_7$ : C, 55.39; H, 5.04; N, 2.69. Found: C, 55.43; H, 5.02; N, 2.72.

**Diethyl 2-acetyl-8-bromo-3-(cyclohexylamino)-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2q):** orange solid (393 mg, 72%); mp 152–153 °C; IR (KBr)  $\nu_{\max}$  3429, 1724, 1712, 1688, 1647  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.43–2.27 (10H, m), 2.75 (3H, s), 3.49–3.56 (1H, m), 4.13–4.23 (4H, m), 7.31 (1H, d,  $J = 8.4 \text{ Hz}$ ), 7.70 (1H, dd,  $J = 8.4, 2.4 \text{ Hz}$ ), 9.60 (1H, d,  $J = 2.4 \text{ Hz}$ ), 8.44 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 23.7, 25.1, 26.3, 26.4, 30.0, 30.1, 51.4, 59.2, 61.4, 95.0, 107.4, 118.1, 118.7, 121.1, 128.4, 129.1, 132.2, 146.1, 150.0, 155.3, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 547 ( $M^+ + 2$  [81Br], 13), 545 (M + [79Br], 13), 505 (25), 475 (63), 459 (100), 56 (55). Anal. Calcd for  $C_{26}H_{28}\text{BrNO}_7$ : C, 57.15; H, 5.17; N, 2.56. Found: C, 57.13; H, 5.12; N, 2.54.

**Diethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-8-bromo-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2r):** orange solid (357 mg, 62%); mp 161–163 °C; IR (KBr)  $\nu_{\max}$  3459, 1742, 1721, 1688, 1640  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.06 (9H, s), 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.65 (6H, s), 1.78 (2H, s), 2.74 (3H, s), 7.32 (1H, d,  $J = 8.4 \text{ Hz}$ ), 7.70 (1H, dd,  $J = 8.4, 2.4 \text{ Hz}$ ), 9.59 (1H, d,  $J = 2.4 \text{ Hz}$ ), 8.45 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 27.9, 31.0, 31.6, 31.7, 41.4, 51.2, 59.2, 61.4, 95.3, 107.4, 118.1, 118.7, 121.1, 128.4, 129.1, 132.2, 146.1, 150.0, 155.3, 160.6, 165.0, 167.1, 195.3 ppm;  $m/z$  (EI, 70 eV) 577 ( $M^+ + 2$  [81Br], 16), 575 (M + [79Br], 16), 505 (57), 475 (48), 388 (100), 56 (45). Anal. Calcd for  $C_{28}H_{34}\text{BrNO}_7$ : C, 58.34; H, 5.94; N, 2.43. Found: C, 58.36; H, 5.96; N, 2.43.

**Diethyl 3-(tert-butylamino)-2-acetyl-8-nitro-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2s):** yellow solid (267 mg, 55%); mp 165 °C; IR (KBr)  $\nu_{\max}$  3404, 1710, 1698, 1632, 1603  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.66 (9H, s), 2.74 (3H, s), 4.13–4.26 (4H, m), 7.55 (1H, d,  $J = 9.6 \text{ Hz}$ ), 8.52–8.61 (2H, m), 10.38 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 30.6, 51.1, 61.2, 63.3, 96.2, 107.5, 118.7, 121.1, 121.9, 125.4, 128.1, 144.0, 146.0, 153.1, 155.6, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 486 (19,  $M^+$ ), 444 (49), 416 (55), 384 (100), 56 (50). Anal. Calcd for  $C_{24}H_{26}\text{N}_2\text{O}_9$ : C, 59.25; H, 5.39; N, 5.76. Found: C, 59.29; H, 5.36; N, 5.75%.

**Diethyl 2-acetyl-3-(cyclohexylamino)-8-nitro-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (3t):** yellow solid (286 mg, 56%); mp 173–175 °C; IR (KBr)  $\nu_{\max}$  3420, 1712, 1701, 1638, 1603  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.43–2.27 (10H, m), 2.75 (3H, s), 3.49–3.56 (1H, m), 4.13–4.23 (4H, m), 7.55 (1H, d,  $J = 9.6 \text{ Hz}$ ), 8.52–8.61 (2H, m), 10.39 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 23.7, 25.1, 26.3, 26.4, 30.0, 30.1, 51.4, 61.3, 63.3, 96.2, 107.5, 118.7, 121.1, 121.9, 125.4, 128.1, 144.1, 146.1, 153.1, 155.3, 160.6, 165.1, 167.1, 195.6 ppm;  $m/z$  (EI, 70 eV) 486 (12,  $M^+$ ), 442 (38), 410 (57), 313 (100), 56 (49). Anal. Calcd for  $C_{26}H_{28}\text{N}_2\text{O}_9$ : C, 60.93; H, 5.51; N, 5.47. Found: C, 60.86; H, 5.50; N, 5.48.

**Diethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-8-nitro-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2u):** yellow solid (271 mg, 50%); mp 163–165 °C; IR (KBr)  $\nu_{\max}$  3429, 1722, 1710, 1647, 1610  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.05 (9H, s), 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.65 (6H, s), 1.78 (2H, s), 2.74 (3H, s), 4.13–4.26 (4H, m), 7.55 (1H, d,  $J = 9.6 \text{ Hz}$ ),

8.52–8.61 (2H, m), 10.39 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 27.9, 31.0, 31.6, 31.7, 43.4, 51.2, 61.4, 63.3, 96.9, 107.6, 118.7, 121.1, 121.9, 125.4, 128.1, 144.1, 146.1, 153.1, 155.3, 160.6, 165.1, 167.1, 195.4 ppm;  $m/z$  (EI, 70 eV) 542 (11,  $M^+$ ), 472 (35), 440 (65), 313 (100), 56 (55). Anal. Calcd for  $C_{28}H_{34}\text{N}_2\text{O}_9$ : C, 61.98; H, 6.32; N, 5.19. Found: C, 62.02; H, 6.36; N, 5.15.

**Diethyl 3-(tert-butylamino)-2-acetyl-6-methoxy-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2v):** yellow solid (259 mg, 55%); mp 153–155 °C; IR (KBr)  $\nu_{\max}$  3404, 1712, 1701, 1648, 1610  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.66 (9H, s), 2.74 (3H, s), 3.93 (3H, s), 4.13–4.26 (4H, m), 7.27 (1H, d,  $J = 8.4, 2.4 \text{ Hz}$ ), 8.63 (1H, d,  $J = 2.4 \text{ Hz}$ ), 9.28 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 30.6, 55.7, 59.1, 61.4, 63.3, 97.5, 107.5, 113.5, 119.1, 120.1, 125.4, 128.1, 142.1, 147.0, 148.0, 155.3, 160.6, 165.0, 167.0, 195.0 ppm;  $m/z$  (EI, 70 eV) 471 (20,  $M^+$ ), 457 (56), 355 (67), 268 (100), 56 (55). Anal. Calcd for  $C_{25}H_{29}\text{NO}_8$ : C, 63.68; H, 6.20; N, 2.97. Found: C, 63.75; H, 6.20; N, 3.02.

**Diethyl 2-acetyl-3-(cyclohexylamino)-6-methoxy-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2w):** yellow solid (283 mg, 57%); mp 158–160 °C; IR (KBr)  $\nu_{\max}$  3420, 1712, 1706, 1657, 1610  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.43–2.27 (10H, m), 2.75 (3H, s), 3.49–3.56 (1H, m), 3.93 (3H, s), 4.13–4.26 (4H, m), 7.27 (1H, d,  $J = 8.4 \text{ Hz}$ ), 7.71 (1H, dd,  $J = 8.4, 2.4 \text{ Hz}$ ), 8.63 (1H, d,  $J = 2.4 \text{ Hz}$ ), 9.28 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 23.7, 25.1, 26.3, 26.4, 30.0, 30.1, 51.7, 55.7, 61.4, 63.3, 97.1, 107.4, 113.5, 119.1, 120.1, 125.5, 128.1, 142.1, 147.0, 148.0, 155.4, 160.6, 165.1, 167.1, 195.3 ppm;  $m/z$  (EI, 70 eV) 497 (17,  $M^+$ ), 483 (50), 453 (25), 268 (100), 56 (55). Anal. Calcd for  $C_{27}H_{31}\text{NO}_8$ : C, 65.18; H, 6.28; N, 2.82. Found: C, 65.27; H, 6.27; N, 2.84.

**Diethyl 3-(2,4,4-trimethylpentan-2-ylamino)-2-acetyl-6-methoxy-5-oxacyclopenta[a]naphthalen-4(2H)-one-1,2-dicarboxylate (2x):** yellow solid (284 mg, 54%); mp 149–151 °C; IR (KBr)  $\nu_{\max}$  3429, 1714, 1705, 1648, 1610  $\text{cm}^{-1}$ ;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ) 1.05 (9H, s), 1.17 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.30 (3H, t,  $J = 7.2 \text{ Hz}$ ), 1.65 (6H, s), 1.78 (2H, s), 2.74 (3H, s), 3.93 (3H, s), 4.13–4.26 (4H, m), 7.27 (1H, d,  $J = 8.4 \text{ Hz}$ ), 7.71 (1H, dd,  $J = 8.4, 2.4 \text{ Hz}$ ), 8.63 (1H, d,  $J = 2.4 \text{ Hz}$ ), 9.27 (1H, s) ppm;  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 13.9, 14.6, 25.9, 27.9, 31.0, 31.6, 31.7, 45.0, 51.7, 55.7, 61.4, 63.3, 97.1, 107.4, 113.4, 119.2, 120.1, 125.5, 128.1, 142.2, 147.0, 148.0, 155.3, 160.2, 165.1, 167.3, 195.2 ppm;  $m/z$  (EI, 70 eV) 527 (19,  $M^+$ ), 513 (34), 499 (45), 268 (100), 56 (55). Anal. Calcd for  $C_{29}H_{37}\text{NO}_8$ : C, 66.02; H, 7.07; N, 2.65. Found: C, 66.31; H, 7.15; N, 2.58.

## ASSOCIATED CONTENT

### Supporting Information

<sup>1</sup>H and <sup>13</sup>C NMR spectra of all compounds; X-ray data for compound **2q** (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

## AUTHOR INFORMATION

### Corresponding Author

\*E-mail: ghandi@khayam.ut.ac.ir.

### Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

We acknowledge the University of Tehran for financial support of this research.

## REFERENCES

- (a) Zhu, J.; Bienamye, H. *Multicomponent Reactions*; VCH: Weinheim, 2005. (b) Simon, C.; Constantieux, T.; Rodriguez, J. *Eur. J. Org. Chem.* **2004**, 4957–4980. (c) Murakami, M. *Angew. Chem., Int. Ed.* **2003**, 42, 718–720. Jacobi Von Vangelin, A.; Neumann, H.; Grdes, D.; Klaus, S.; Strübing, D.; Beller, M. *Chem.—Eur. J.* **2003**, 9, 4286–

4294. (d) Balme, G.; Bossharth, E.; Monterio, N. *Eur. J. Org. Chem.* **2003**, 4101–4111.
- (2) (a) Dömling, A. *Chem. Rev.* **2006**, *106*, 17–89. (b) Hulme, C.; Gore, V. *Curr. Med. Chem.* **2003**, *10*, 51–80. (c) Nair, V.; Rajesh, C.; Vinod, A. U.; Bindu, S.; Sreekanth, A. R.; Mathen, J. S.; Balagopal, L. *Acc. Chem. Res.* **2003**, *36*, 899–907. (d) Ugi, I.; Verner, B.; Dmling, A. *Molecules* **2003**, *8*, 53–66. (e) Zhu, J. *Eur. J. Org. Chem.* **2003**, 1133–1144.
- (3) (a) Zhou, L. H.; Su, G.; Zhang, W.; Yan, B. *J. Comb. Chem.* **2009**, *11*, 1083–1093. (b) Lu, K.; Luo, T.; Xiang, Z.; You, Z.; Fathi, R.; Chen, J.; Yang, Z. *J. Comb. Chem.* **2005**, *7*, 958–967. (c) Zhu, J. *Eur. J. Org. Chem.* **2003**, 1133–1144. (d) Weber, L. *Curr. Med. Chem.* **2002**, *9*, 1241–1253. (e) Bienayme, H.; Hulme, C.; Oddon, G.; Schmidt, P. *Chem.—Eur. J.* **2000**, *6*, 3321–3329. (f) R. Orru, R. V. A.; de Greef, M. *Synthesis* **2003**, 1471–1499. (g) Dömling, A.; Ugi, I. *Angew. Chem., Int. Ed.* **2000**, *39*, 3168–3210. (h) Lee, D.; Sello, J. K.; Schreiber, S. L. *Org. Lett.* **2000**, *2*, 709–712. (i) Armstrong, R. W.; Combs, A. P.; Tempest, P. A.; Brown, A. D.; Thomas, A. K. *Acc. Chem. Res.* **1996**, *29*, 123–131.
- (4) (a) Hall, D. G.; Manku, S.; Wang, F. *J. Comb. Chem.* **2001**, *3*, 125–150. (b) Nicolaou, K. C.; Pfefferkorn, J. A.; Barluenga, S.; Mitchell, H. J.; Roecker, A. J.; Cao, G.-Q. *J. Am. Chem. Soc.* **2000**, *122*, 9968–9976. (c) Wipf, P.; Reeves, J. T.; Balachandran, R.; Giuliano, K. A.; Hamel, E.; Day, B. W. *J. Am. Chem. Soc.* **2000**, *122*, 9391–9395. (d) Boger, D. L.; Fink, B. E.; Hedrick, M. P. *J. Am. Chem. Soc.* **2000**, *122*, 6382–6394. (e) Lee, K. J.; Angulo, A.; Ghazal, P.; Janda, K. D. *Org. Lett.* **1999**, *1*, 1859–1862. (f) Xu, R.; Greiveldinger, G.; Marenus, L. E.; Cooper, A.; Ellman, J. A. *J. Am. Chem. Soc.* **1999**, *121*, 4898–4899. (g) Nicolaou, K. C.; Winssinger; Vourloumis, D. *T. J. Am. Chem. Soc.* **1998**, *120*, 10814–10826. (h) Ugi, I. *Angew. Chem., Int. Ed.* **1962**, *1*, 8–20.
- (5) Winterfeldt, E.; Schumann; Dillinger, H. *J. Chem. Ber.* **1969**, *102*, 1656–1664.
- (6) Stachulski, A. V.; Berry, N. G.; Low, A. C. L.; Moores, S. L.; Row, E.; Warhurst, D. C.; Adagu, I. S.; Rossignol, J.-F. *J. Med. Chem.* **2006**, *49*, 1450–1454.
- (7) Nawrot-Modranka, J.; Nawrot, E.; Graczyk, J. *Eur. J. Med. Chem.* **2006**, *41*, 1301–1309.
- (8) (a) Zarganes-Tzitzikas, T.; Terzidis, M. A.; Stephanidou-Stephanatou, J.; Tsoleridis, C. A.; Kostakis, J. E. *J. Org. Chem.* **2011**, *76*, 9008–9014. (b) Terzidis, M. A.; Stephanidou-Stephanatou, J.; Tsoleridis, C. A. *J. Org. Chem.* **2010**, *75*, 1948–1955.
- (9) (a) Nawrot-Modranka, J.; Nawrot, E.; Graczyk, J. *Eur. J. Med. Chem.* **2006**, *41*, 1301–1309. (b) Kirkiacharian, S.; Thuy, D. T.; Sicic, S.; Bakhchinian, R.; Kurkjian, R.; Tonnaire, T. *Farmac.* **2002**, *57*, 703–708. (c) Mao, P.C.-M.; Mouscadet, J. F.; Leh, H.; Auclair, C.; Hsu, L. Y. *Chem. Pharm. Bull. (Tokyo)* **2002**, *50*, 1634–1637 (c).
- (10) (a) Kawase, M.; Varu, B.; Shah, A.; Motohashi, N.; Tani, S.; Saito, S.; Debnath, S.; Mahapatra, S.; Dastidar, S. G.; Chakrabarty, A. N. *Arzneim.-Forsch./Drug Res.* **2001**, *51*, 67–71. (b) Kayser, O.; Kolodziej, H. Z. *Naturforsch.* **1999**, *54*, 169–174.
- (11) (a) Jimenez-Orozco, F. A.; Lopez-Gonzalez, J. S.; Nieto-Rodriguez, A.; Velasco-Velazquez, M. A.; Molina-Guarneros, J. A.; Mendoza-Patino, N.; Garcia-Mondragon, M. J.; Elizalde-Galvan, P.; Leon-Cedeno, F.; Mandoki, J. *J. Lung Cancer* **2001**, *34*, 185–194. (b) Valenti, P.; Fabbri, G.; Rampa, A.; Bisi, A.; Gobbi, S.; Da Re, P.; Carrara, M.; Sgevano, A.; Cima, L. *Anti-Cancer Drug Des.* **1996**, *11*, 243–252.
- (12) Kamat, D. P.; Tilve, S. G.; Kamat, V. P. *Tetrahedron Lett.* **2012**, *53*, 4469–4472.
- (13) Neel, M.; Gouin, J.; Voituriez, A.; Marinett, A. *Synthesis* **2011**, 2003–2009.
- (14) (a) Ghandi, M.; Zarezadeh, N.; Taheri, A. *Tetrahedron Lett.* **2012**, *53*, 3353–3356. (b) Ghandi, M.; Zarezadeh, N.; Taheri, A. *Tetrahedron Lett.* **2011**, *52*, 1228–1232. (c) Ghandi, M.; Zarezadeh, N.; Taheri, A. *Tetrahedron* **2010**, *66*, 8231–8237.
- (15) (a) Nair, V.; Vinod, A. U.; Abhilash, N.; Menon, R. S.; Santhi, V.; Varma, R. L.; Viji, S.; Mathew, S.; Srinivas, R. *Tetrahedron* **2003**, *59*, 10279–10286. (b) Junjappa, H.; Saxena, M. K.; Ramaiah, D.; Loharay, B. B.; Rath, N. P.; George, M. V. *J. Org. Chem.* **1998**, *63*, 9801–9805.
- (c) Dillinger, H. J.; Fengler, G.; Schumann, D.; Winterfeldt, E. *Tetrahedron* **1974**, *30*, 2553–2559. (d) Oakes, T. R.; David, H. G.; Nagel, F. *J. Am. Chem. Soc.* **1969**, *91*, 4761–4765. (e) Winterfeldt, E.; Giesler, G. *Chem. Ber.* **1968**, *101*, 4022–4031.
- (16) The CCDC deposition number for compound **2q** is 915923. Formula:  $C_{26}H_{28}BrNO_7$ . Unit cell parameters:  $a = 9.5443(6)$  Å,  $b = 10.3293(5)$  Å,  $c = 13.6741(8)$  Å,  $\alpha = 93.636(4)$ °,  $\beta = 98.925(5)$ °,  $\gamma = 113.086(5)$ °, space group *P-1*.
- (17) Carey, F. A.; Sundberg, R. J. *Advanced Organic Chemistry, Part A: Structure and Mechanisms*, 5th ed.; Springer: New York, 2007; 344.